Name:	Date:			HURBAN
Chemistry ~ Ms. Hart	<u>Class:</u>	Anions or	Cations	SCHOOL FOR CRIMINAL
<u>8.1</u>	<u>o Exit Ticket</u>			5
1. What is the effect of press	ure on volume?	(use the trend/re	lationship pos	ster on the shades to help!)

2. A gas occupies 11.2 liters at 0.860 atm. What is the pressure if the volume becomes 15.0 L?

Name:	Date:			HURBAN HASSEMBLY	
Chemistry ~ Ms. Hart	<u>Class:</u>	Anions	or	Cations	SCHOOL FOR CRIMINAL IUSTICE
8.10	Exit Ticket**				5

1. 600.0 mL of a gas is at a pressure of 8.00 atm. What is the volume of the gas at 2.00 atm?

2. What is the effect of pressure on volume? (use the trend/relationship poster on the shades to help!)