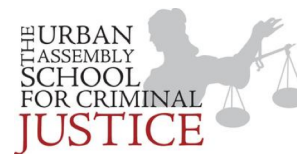


Name: _____ Date: _____

Chemistry ~ Ms. Hart

Class: Anions or Cations



8.10 Exit Ticket

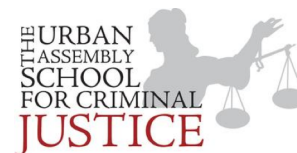
1. What is the effect of pressure on volume? (use the trend/relationship poster on the shades to help!)

2. A gas occupies 11.2 liters at 0.860 atm. What is the pressure if the volume becomes 15.0 L?

Name: _____ Date: _____

Chemistry ~ Ms. Hart

Class: Anions or Cations



8.10 Exit Ticket**

1. 600.0 mL of a gas is at a pressure of 8.00 atm. What is the volume of the gas at 2.00 atm?

2. What is the effect of pressure on volume? (use the trend/relationship poster on the shades to help!)
