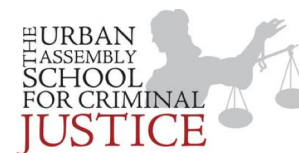


Name: _____ Date: _____

Chemistry ~ Ms. Hart

Class:

Anions or Cations



5.11 Exit Ticket

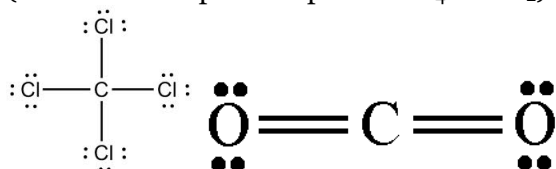
1) The bond between hydrogen and oxygen in a water molecule is classified as

1. ionic and nonpolar
2. ionic and polar
3. covalent and nonpolar
4. covalent and polar
- 5.

2) Which bond has the most ionic character?

1. H—F
2. H—Cl
3. H—Br
4. H—I

3) How can a bond in a molecule be polar while the overall molecule is non-polar?
(Use this example to explain: CCl_4 or CO_2). Use the back if necessary!

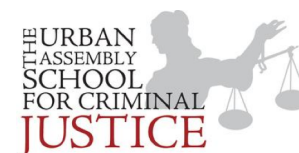


Name: _____ Date: _____

Chemistry ~ Ms. Hart

Class:

Anions or Cations



5.11 Exit Ticket

1) The bond between hydrogen and oxygen in a water molecule is classified as

6. ionic and nonpolar
7. ionic and polar
8. covalent and nonpolar
9. covalent and polar
- 10.

2) Which bond has the most ionic character?

5. H—F
6. H—Cl
7. H—Br
8. H—I

3) How can a bond in a molecule be polar while the overall molecule is non-polar?
(Use this example to explain: CCl_4 or CO_2). Use the back if necessary!

